

Steady-state v-distribution in two layers; temperature differences

(only yellow cells may be changed!)

$$Z_A = 100 \cdot 10^3 \text{ s/m}$$

$$Z_B = 50 \cdot 10^3 \text{ s/m}$$

Use vapour contents!

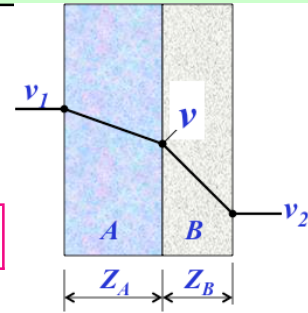
$$v_1 = 10.0 \text{ g/m}^3$$

$$v_2 = 5.0 \text{ g/m}^3$$

gives $v = 6.7 \text{ g/m}^3$

$$v = v_1 - \frac{Z_A}{Z_A + Z_B} (v_1 - v_2)$$

NOTE! Check that it is not larger than $v_s(T)$!



Steady-state RH-distribution in two layers; constant temperature

(only yellow cells may be changed!)

$$Z_A = 100 \cdot 10^3 \text{ s/m}$$

$$Z_B = 50 \cdot 10^3 \text{ s/m}$$

OK to use RH!

$$RH_1 = 100 \%$$

$$RH_2 = 50 \%$$

gives $RH = 66.7 \%$

$$RF = RF_1 - \frac{Z_A}{Z_A + Z_B} (RF_1 - RF_2)$$

"RF" in equation and figure means "RH"!

